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Physical Science Reading and Study Workbook Level B Chapter 6 61 IPLS Section 6.2 Covalent Bonding (pages 165–169) This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules. Reading Strategy(page 165) Relating Text and Visuals As you read the section, look closely at Figure 9 ...

### **Chapter 6 Chemical Bonds Section 6.2 Covalent Bonding**

1 Section 6-2 Covalent Bonding Molecule A neutral group of atoms held together by covalent bonds. A single molecule is an individual unit that can exist on its own.

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### **SECTION 6 2 COVALENT BONDING ANSWER KEY PDF**

Unequal Sharing of Electrons (pages 167–168) 6. In general, elements at the of a group have a greater attraction for electrons than elements at the

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